

LA-UR-15-22380 (Accepted Manuscript)

Distribution of neptunium and plutonium in New Mexico lichen samples (Usnea arizonica) contaminated by atmospheric fallout

Oldham, Warren James Hanson, Susan Kloek Lavelle, Kevin B. Miller, Jeffrey L.

Provided by the author(s) and the Los Alamos National Laboratory (0000-00-00).

To be published in: Journal of Radioanalytical and Nuclear Chemistry ; 30 August 2015

DOI to publisher's version: 10.1007/s10967-015-4402-0

Permalink to record: http://permalink.lanl.gov/object/view?what=info:lanl-repo/lareport/LA-UR-15-22380

Disclaimer:

Approved for public release. Los Alamos National Laboratory, an affirmative action/equal opportunity employer, is operated by the Los Alamos National Security, LLC for the National Nuclear Security Administration of the U.S. Department of Energy under contract DE-AC52-06NA25396. Los Alamos National Laboratory strongly supports academic freedom and a researcher's right to publish; as an institution, however, the Laboratory does not endorse the viewpoint of a publication or guarantee its technical correctness.



- 1 Special Issue (SI): MARC X
- 2 LOG NUMBER OF PAPER: 335
- 3 TITLE OF PAPER: Distribution of neptunium and plutonium in New Mexico lichen
- 4 samples (Usnea arizonica) contaminated by atmospheric fallout
- 5 AUTHORS: Warren J Oldham, Jr.¹, Susan K. Hanson¹, Kevin B. Lavelle², Jeffrey L.
- 6 Miller¹
- 7 POSTAL ADDRESS OF AUTHORS: (1) Nuclear and Radiochemistry Group (C-NR),
- 8 P.O. Box 1663, Los Alamos National Laboratory, Los Alamos, New Mexico 87545, USA
- 9 (2) Department of Chemistry, University of Cincinnati, PO Box 210172, Cincinnati,
- 10 OH 45221-0172, USA
- 11 CORRESPONDING AUTHOR: woldham@lanl.gov, telephone: (505) 667-5900, fax
 12 (505) 606-1801
- 13

14

Title page

- Names of the authors: Warren J. Oldham, Jr.¹, Susan K. Hanson¹, Kevin B. Lavelle², and
 Jeffrey L. Miller¹
- 17 Title: Distribution of neptunium and plutonium in New Mexico lichen samples
 18 (Usnea arizonica) contaminated by atmospheric fallout
- 19 Affiliation(s) and address(es) of the author(s): (1) Nuclear and Radiochemistry Group (C-
- 20 NR), P.O. Box 1663, Los Alamos National Laboratory, Los Alamos, New Mexico 87545,
- 21 USA (2) Department of Chemistry, University of Cincinnati, PO Box 210172, Cincinnati,
- 22 OH 45221-0172, USA
- 23 E-mail address of the corresponding author: woldham@lanl.gov
- 24

Distribution of neptunium and plutonium in New Mexico lichen samples (Usnea arizonica) contaminated by atmospheric fallout

28 Warren J. Oldham, Jr.¹, Susan K. Hanson¹, Kevin B. Lavelle², and Jeffrey L. Miller¹

¹ Nuclear and Radiochemistry Group (C-NR), P.O. Box 1663, Los Alamos National
 Laboratory, Los Alamos, New Mexico 87545, USA

² Department of Chemistry, University of Cincinnati, PO Box 210172, Cincinnati,
 OH 45221-0172, USA

33 Abstract

The concentrations of ²³⁷Np, ²³⁹Pu and ²⁴⁰Pu were determined in lichen samples (Usnea 34 35 arizonica) that were collected from ten locations in New Mexico between 2011 and 2013 36 using isotope dilution inductively-coupled plasma mass spectrometry (ID-ICP-MS). The observed isotopic ratios for ²³⁷Np/²³⁹Pu and ²⁴⁰Pu/²³⁹Pu indicate trace contamination from 37 38 global and regional fallout (e.g. Trinity test and atmospheric testing at the Nevada Test 39 Site). The fact that actinide contamination is detected in recent lichen collections suggests 40 continuous re-suspension of fallout radionuclides even 50 years after ratification of the 41 Limited Test Ban Treaty.

42 Keywords

43 Neptunium-237, Plutonium isotopes, atmospheric fallout, ICP-MS, lichen

44 Introduction

45 Lichens obtain essential nutrients directly through atmospheric deposition and have 46 evolved highly efficient mechanisms to bioconcentrate trace elements within their tissues 47 [1]. This characteristic has been used for many years in both Europe and North America 48 to monitor the distribution of atmospheric pollutants [2,3]. With respect to the Four 49 Corners region of the United States, the pattern of trace, minor, and earth abundant 50 elements measured in the epilithic lichen *Xanthoparmelia* spp. was used to distinguish 51 natural and anthropogenic emissions (agriculture, mining, industrial activities and urban 52 traffic) [4]. This same species was employed by Thomas and Ibrahim to characterize the 53 distribution of plutonium surrounding the Rocky Flats nuclear facility in Colorado [5].

54 In this paper the lichen Usnea spp. was evaluated as a potential biomonitor for trace 55 transuranic contamination. Usnea is a widely distributed, yellowish-green fruticose genus 56 of lichen [6]. Occurring within montane regions of Arizona, New Mexico and Colorado, 57 the epiphytic species Usnea arizonica (western bushy beard) typically grows on 58 ponderosa and piñon pine trees [7]. Because this species grows several meters above the 59 ground surface, the proposed actinide measurements may reflect regional atmospheric 60 transport (resuspension) rather than superficial contamination from adjacent soils. Studies 61 of this type could find utility in environmental monitoring programs associated with 62 modern nuclear activities [8]. The present work was undertaken to define background concentrations of ²³⁷Np, ²³⁹Pu, and ²⁴⁰Pu present in samples of Usnea arizonica retrieved 63 64 from remote locations in New Mexico, USA.

65 **Experimental**

66 *Usnea Lichen Collections* The lichen samples were acquired between June 2011 and 67 November 2013 from forested mountainous regions in New Mexico (Table 1). In general 68 the preferred habitat of *Usnea arizonica* occurs above 7500 feet in piñon and ponderosa 69 pine forests. However, a sample was also collected from White Rock, NM where the 70 lower elevation corresponds to a generally hotter and drier climate less favorable to this 71 species. At this site *Usnea* sp. was located on piñon trees that were partially protected 72 from the sun on the south side of an east/west trending canyon. The dry mass submitted

for analysis varied from ~8 to 29.5 g. Samples were dried, a dry mass recorded, and then ashed at 550° C to provide from ~0.6 to 1.8 g of inorganic residue. The ash was readily dissolved upon repeated fuming with HNO₃/HF in a Teflon beaker. After evaporating the acid mixture to dryness a stock solution was prepared with 3M HCl to which was added H₃BO₃ to scavenge fluoride and to facilitate re-dissolution of insoluble CaF₂ and MgF₂ formed in the HF fuming process.

Location	Collection Date (mo-yr)	Location (lat; lon)	Elev. (feet)	Dry lichen mass (g)	Lichen ash mass (g)
Bluewater	Jun-11	35° 15.939' N; 108° 7.089' W	7509	8.1726	0.5983
Bluewater	May-13	35° 15.939' N; 108° 7.089' W	7509	13.7708	1.0964
Gallinas Peak	Aug-11	34° 14.815' N; 105° 47.309' W	8633	12.6920	0.8307
Gallinas Peak	Aug-11	34° 14.815' N; 105° 47.309' W	8633	9.0595	0.6534
Seven Springs	Aug-13	35° 55.500' N; 106° 42.281' W	7940	18.0553	0.7909
Norski XC	Oct-13	35° 47.409' N; 105° 48.693' W	10340	17.3440	1.1209
Pajarito Ski Area	Oct-13	35° 53.685' N; 106° 23.853' W	9510	20.0424	1.3597
Quemado Lake	Oct-13	34° 7.793' N; 108° 29.878' W	7899	21.3580	0.8775
Apache NF	Oct-13	34° 7.808' N; 108° 26.104' W	8007	26.8604	1.8111
Carson NF	Oct-13	36° 25.270' N; 105° 20.600' W	8986	21.7681	0.9273
White Rock	Nov-13	35° 49.038' N; 106° 12.743' W	6435	29.4687	1.5636
Williams Lake	Nov-13	36° 34.003' N; 105° 25.931' W	10800	23.3811	1.1572

79 **Table 1** Lichen collection detail

80 ICP-MS instrumentation and operating conditions Purified samples (vide infra) 81 dissolved in 2% HNO₃ were analyzed using a Thermo X-series II quadrupole ICP-MS equipped with an ESI APEX IR sample inlet system. This instrumental configuration 82 provides routine measurement sensitivity of $\sim 3 \times 10^6$ cps/ppb (²³⁸U). The detector dead 83 84 time was optimized according to the procedure of Vanhaecke [9] to ensure consistent isotopic ratio measurements. Instrumental mass bias was monitored using the isotopic 85 86 standards, NBL U500 and CRM 128. The instrumental count rates at mass 237, 239 and 240 amu were measured relative to an internal ²⁴²Pu (NIST SRM 4334G) spike to 87 88 determine absolute isotopic concentrations and the count rate at mass 238 was monitored to correct for minor interference at mass 239 due to ²³⁸U¹H. Results were calculated from 89 90 the average of 8 replicate analyses. Each replicate (peak hopping mode) represents the 91 average count rate for 800 sweeps (30 ms dwell time for all monitored isotopes). Measurement uncertainty is expressed as 1-sigma standard deviation for the 8 replicate 92 analyses. The uncertainty in absolute ²³⁹Pu concentration includes the NIST certified 93

94 error in ²⁴²Pu spike concentration. Further details of the operating conditions and

95 instrument performance are provided in Table 2.

Parameters	ICP-MS Xseries II
Power	1400 W
Gas flows	Cool gas: 13 L/min
	Auxiliary gas: 0.60-0.65 L/min
	Nebulizer gas: 0.74-0.78 L/min
Sensitivity (²³⁸ U)	$3 \times 10^6 \text{ cps/ppb}$
Backgrounds (2% HNO ₃)	<0.5 cps
Oxides (Ce) and double charge ions (Ba)	< 3%
Sample Inlet System	ESI APEX IR
Spray and flow rate	Self-aspirating PFA nebulizer: 0.28 mL/min
Cones	Ni sample and skimmer cones (Xs)
Standard resolution	0.75 amu (10% of peak height)
$^{238}\text{U}^{1}\text{H}^{238}\text{U}$	3×10^{-5}

96 Table 2 Analytical parameters and settings of ICP-MS Xseries II

98 *Purification procedure* To the dissolved sample was added a known quantity of 242 Pu (0.7-1.2 x 10¹² atoms) tracer that was previously prepared from NIST SRM 4334G. 99 A purified ²³⁷Np/Pu sample was prepared for ICP-MS assay by first pre-concentrating the 100 101 transuranium elements using a LaF₃ precipitation step, followed by anion exchange 102 column chromatography. Details of column preparation, valence adjustment and wash volumes have been previously reported [10]. Minor fractionation of ²³⁷Np and Pu that 103 104 occurs during the purification process was corrected by measuring the fractionation of an in-house 237 Np/ 242 Pu standard that is analyzed in parallel with the environmental samples 105 106 [10]. Confidence in the analytical procedure was provided through measurements of acid 107 dissolution blanks and laboratory process blanks that were processed in parallel with the lichen samples. From these data an estimated method detection limit of $\sim 5 \times 10^6$ atoms 108 109 was established. In addition the procedure was used to analyze aliquots of a stock solution of dissolved NIST SRM 4357 (natural matrix radioactivity standard) for ²³⁷Np, 110 ²³⁹Pu, and ²⁴⁰Pu isotopes. Aliquots of SRM 4357 containing ~0.27 g of dissolved 111 sediment (~1.6 x 10⁹ atoms of ²³⁹Pu) were selected for analysis alongside the 112 113 environmental samples to mimic the concentrations of transuranic isotopes expected in 114 the lichen samples. These quality control results are presented in Table 3 and compare well with previous reports (see [10] and references therein). 115

⁹⁷

Sample ID	²³⁹ Pu (atoms/g sediment)	²⁴⁰ Pu/ ²³⁹ Pu	²³⁷ Np/ ²³⁹ Pu
4357-01	5.97(10) x 10 ⁹	0.2369 ± 0.0080	0.1512 ± 0.0095
4357-02	$5.97(9) \ge 10^9$	0.2360 ± 0.0093	0.1508 ± 0.0085
4357-03	$6.02(10) \ge 10^9$	0.2341 ± 0.0092	0.1550 ± 0.0080
4357-04	$6.02(10) \ge 10^9$	0.2285 ± 0.0040	0.1469 ± 0.0069
4357-05	$5.89(9) \ge 10^9$	0.2339 ± 0.0086	0.1478 ± 0.0071
4357-06	$5.94(9) \ge 10^9$	0.2315 ± 0.0041	0.1561 ± 0.0078
Average $\pm 2\sigma$	5.97(10) x 10 ⁹	0.2335 ± 0.0061	0.1513 ± 0.0075

116	Table 3 Analyt	ical results c	obtained for	~0.27 g a	aliquots o	of NIST SRM 4357

117

118 **Results**

The analytical results of this study are summarized in Table 4. The concentration of ²³⁹Pu measured for each of the *Usnea* spp. lichen samples is presented in units of atoms per gram of lichen ash. This particular convention was chosen to simplify comparison with typical soil analyses. If desired, these data can be expressed in units of atoms per gram of dried lichen using the mass information recorded in Table 1. The transuranic isotopic composition is reported as atom ratios for ²⁴⁰Pu/²³⁹Pu and ²³⁷Np/²³⁹Pu with associated 1sigma standard deviation measurement uncertainty.

. /	Mexico <i>Ushea</i> spp. ne	men samples	²³⁹ Pu		
	Location	Location Sample ID		²⁴⁰ Pu/ ²³⁹ Pu	²³⁷ Np/ ²³⁹ Pu
	Bluewater	BWC-01	$1.84(3) \ge 10^9$	0.1542 ± 0.0047	Not measured
	Bluewater	BWC-02	1.57(3) x 10 ⁹	0.1347 ± 0.0029	0.1268 ± 0.0064
	Gallinas Peak	GP-01	2.42(4) x 10 ⁹	0.0919 ± 0.0028	0.1466 ± 0.0085
	Gallinas Peak	GP-02	$1.27(3) \ge 10^9$	0.0569 ± 0.0047	0.0674 ± 0.0071
	Seven Springs	SS-01	$6.39(2) \ge 10^8$	0.1187 ± 0.0068	0.2480 ± 0.0162
	Norski XC	NXC-01	$1.78(3) \ge 10^9$	0.1457 ± 0.0051	0.2057 ± 0.0087
	Pajarito Ski Area	PSA-01	$1.12(2) \times 10^9$	0.1548 ± 0.0033	0.3556 ± 0.0172
	Quemado Lake	QLO-01	$2.12(3) \times 10^9$	0.1491 ± 0.0046	0.2283 ± 0.0110
	Apache NF	ANF-01	$3.15(7) \ge 10^8$	0.1331 ± 0.0090	0.1723 ± 0.0133
	Carson NF	CNF-01	$2.30(7) \ge 10^8$	0.1430 ± 0.0065	0.3685 ± 0.0207
	White Rock	WR-01	$1.18(2) \ge 10^9$	0.1059 ± 0.0052	0.0682 ± 0.0051
	Williams Lake Trail	WLT-01	$2.15(3) \ge 10^9$	0.1637 ± 0.0043	0.2342 ± 0.0107

Table 4 Concentrations of ²³⁹Pu and ²⁴⁰Pu/²³⁹Pu and ²³⁷Np/²³⁹Pu atom ratios for New
 Mexico Usnea spp. lichen samples

128

129

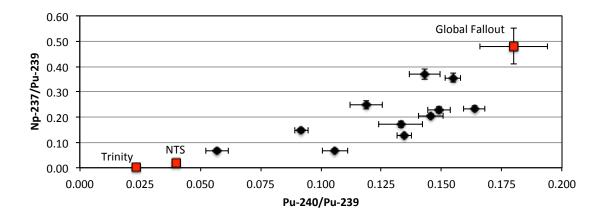
130 **Discussion**

131 A number of potential source terms have contributed transuranic isotopes to the 132 environment of New Mexico. The most prominent is Global Fallout due to large 133 thermonuclear tests carried out by both the United States and former Soviet Union [11]. 134 An important regional contributor, especially in the northern mountain areas of the state, 135 is fallout from low yield nuclear tests conducted at the Nevada Test Site (NTS) from 136 1951 to 1962 [12]. In addition, fallout from the Trinity test (July 1945) extends as a relatively faint plume from ground zero (33° 40.638' N; 106° 28.524' W) to the northeast 137 138 [13]. Though not detected in this study, reactor derived plutonium due to the Chernobyl 139 (1986) and Fukushima (2011) accidents is also considered [14-16]. Each of these sources 140 is characterized by a unique composition that will ultimately define the isotopic pattern 141 measured in a collection of environmental samples.

The absolute concentration of ²³⁹Pu measured in the *Usnea* sp. lichen samples is broadly comparable to soils in the region. The ²³⁹⁺²⁴⁰Pu activity concentrations reported by Purtyman, et al. correspond to a ²³⁹Pu concentration range of 3 x 10⁷ to 2 x 10⁹ atoms/g [17]. The median soil concentration for ²³⁹Pu was ~3 x 10⁸ atoms/g. Compared to these values the concentration of ²³⁹Pu in lichen ash ranged from 2.3 x 10⁸ to 2.4 x 10⁹ atoms/g. The median concentration from this study was 1.4 x 10⁹ atoms/g, suggesting a possible, admittedly subtle, biological enhancement.

149 The isotopic composition of the Usnea sp. lichen samples are presented in Figure 1, 150 compared to the three probable source terms. Global Fallout is the best characterized endmember and is indicated by average ²⁴⁰Pu/²³⁹Pu and ²³⁷Np/²³⁹Pu Northern Hemisphere 151 152 values of 0.180 ± 0.014 and 0.48 ± 0.07 , respectively [18]. The isotopic composition of fallout from NTS is more uncertain, not only because of the diversity of experiments 153 154 [19], but also due to the particular wind and weather conditions at the time of each test [20]. For the purposes of this comparison, a reasonable NTS signature is approximated by 155 240 Pu/ 239 Pu and 237 Np/ 239 Pu values of 0.04 and ~0.02, where only one significant figure is 156 157 justified [18]. Recent measurements of an archived trinitite sample provide an approximation for ²⁴⁰Pu/²³⁹Pu and ²³⁷Np/²³⁹Pu ratios from Trinity fallout of 0.0235 and 158

159 0.0021 [21]. The isotopic composition of the lichen collection forms an approximate mixing line between these fallout end-members. The two samples collected from the 160 161 summit of Gallinas Peak show the strongest contribution from regional fallout. Indeed, 162 Gallinas Peak lies just 57 miles to the northeast of Trinity ground zero, directly under the 163 fallout plume [13]. While the isotopic results for both lichen samples lie on a mixing line 164 between Global Fallout and Trinity fallout, the fact that their individual isotopic 165 compositions are so different from one another (Table 4), suggests the heterogeneous/particulate nature of fallout in the environment. The White Rock lichen 166 sample is unusually low in both ²³⁷Np and ²⁴⁰Pu concentrations probably reflecting a 167 168 local signature from historic emissions from the nearby Los Alamos National Laboratory 169 [22].

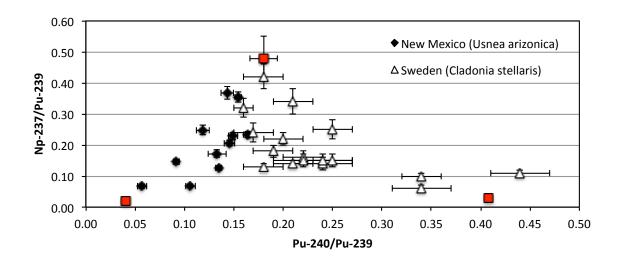


170

Fig. 1 A plot of ²³⁷Np/²³⁹Pu versus ²⁴⁰Pu/²³⁹Pu for *Usnea* sp. lichens collected in New
 Mexico, compared to the isotopic composition of Global Fallout and regional fallout due
 to testing at the Nevada Test Site (NTS) and the Trinity test

The 8 remaining lichen samples (characterized by ${}^{240}Pu/{}^{239}Pu$ ratios ≥ 0.12) were all 174 collected in remote areas of the state that should not be impacted by local nuclear 175 activities. The isotopic inventory for these samples is considered representative of fallout 176 from both Global and NTS sources. For this group the average 240 Pu/ 239 Pu ratio of 0.143 177 ± 0.028 (k = 2) indicates a distinct contribution of transuranium isotopes from low yield 178 tests at NTS, although the ²³⁷Np/²³⁹Pu ratios are scattered such that an idealized mixing 179 line is not observed. Variation in the ²³⁷Np/²³⁹Pu ratios could reflect the diversity of 180 fallout from NTS, but more likely results from inter-element fractionation due to natural 181 weathering and redistribution within the environment. For many of the Usnea sp. lichen 182

samples, the ²³⁷Np concentration tends to be depleted relative to ²³⁹Pu. Similar behavior 183 184 has been reported by Lindahl, et al. for *Cladonia stellaris* lichens in Sweden [14]. The 185 isotopic composition of the Swedish samples reflects mixing of Global Fallout with contamination from the Chernobyl accident (Fig 2). In both the New Mexico and 186 Swedish environments, the ²³⁷Np/²³⁹Pu ratio tends to be lower than expected for an 187 idealized mixing line. A potential explanation is related to the slightly greater 188 environmental (aqueous) mobility of Np compared to Pu [23]. If Np slightly outstrips Pu 189 in downward migration through the soil column, then the ${}^{237}Np/{}^{239}Pu$ ratio in the top 190 191 layer will gradually decline over time [24]. The isotopic composition of transuranic 192 elements measured in lichens is most likely to reflect the uppermost layer of soils that are 193 also the most easily eroded and carried by the wind.



194

195 196

Fig. 2 Comparison of ²³⁷Np/²³⁹Pu and ²⁴⁰Pu/²³⁹Pu ratios for New Mexico *Usnea* spp. lichens with Sweden *Cladonia stellaris* lichens

The concentration of contaminants within the tissues of *Usnea* sp. lichens is assumed to be in equilibrium with the environment [1]. The occurrence of transuranic isotopes in modern lichen collections reflects the background concentration of nuclear fallout that is actively moving through wind erosion and atmospheric transport. These processes serve to redistribute and ultimately homogenize one of the most recognizable signatures of the modern era throughout the surface environment of the Earth on very long timescales [25].

203

204 **Conclusions**

Usnea sp. lichens collected from New Mexico contain transuranic isotopes derived from historic atmospheric nuclear fallout. The concentration of ²³⁷Np, ²³⁹Pu and ²⁴⁰Pu in lichen ash samples is comparable or slightly elevated compared to regional soils. The isotopic composition of the transuranic contamination reflects mixtures of Global Fallout and regional fallout from the Nevada Test Site (NTS) and from the Trinity test. The fact that contamination is detected in recent lichen collections suggests continuous re-suspension of fallout radionuclides even 50 years after ratification of the Limited Test Ban Treaty.

212 Acknowledgements

213 This work was performed under the auspices of the U.S. Department of Energy by Los

214 Alamos National Laboratory under contract DE-AC52-06NA25396. We thank the

215 Laboratory Directed Research and Development Program for financial support (Project

216 Number: 20120459ER). KBL is grateful to the Department of Homeland Security for

210 Rumber 2012043/ER). RDE is grateful to the Department of Homenand Security R

217 financial support as a Nuclear Forensics Graduate Fellow.

218 **References**

1. Nash III TH (2008) Nutrients, elemental accumulation, and mineral cycling. In: Nash
 III TH (ed) Lichen Biology. 2nd edn. Cambridge University Press, New York, p 234

221 2. Richardson DHS, Nieboer E (1981) Lichens and pollution monitoring. Endeavour
222 5:127-133

3. Conti ME, Cecchetti G (2001) Biological monitoring: lichens as bioindicators of air
 pollution assessment - a review. Environmental Pollution 114:471-492

4. Zschau T, Getty S, Gries C, Ameron Y, Zambrano A, Nash III TH (2003) Historical
and current atmospheric deposition to the epilithic lichen *Xanthoparmelia* in Maricopa
County, Arizona. Environmental Pollution 125:21-30

5. Thomas RS, Ibrahim SA (1995) Plutonium concentrations in lichens of Rocky Flats
environs. Health Phys 68 (3):311-319

230 6. Purvis W (2000) Lichens. Smithsonian Institution Press, Washington, D. C.

7. Brodo IM, Sharnoff SD, Sharnoff S (2001) Lichens of North America. Yale University
Press, New Haven

8. Thakur P, Ballard S, Hardy R (2014) Radiation release at the nation's only operating
deep geological repository--an independent monitoring perspective. Environ Sci Technol
48 (21):12698-12705. doi:10.1021/es503649y

9. Vanhaecke F, Wannemacker G, Moens L, Dams R, Latkoczy C, Prohaska T, Stingeder
G (1998) Dependence of detector dead time on analyte mass number in inductively
coupled plasma mass spectrometry. Journal of Analytical Atomic Spectrometry 13:567571

10. Matteson BS, Hanson SK, Miller JL, Oldham WJ, Jr. (2015) Concurrent
determination of Np and Pu isotopes using ICP-MS: analysis of NIST environmental
matrix standard reference materials 4357, 1646a, and 2702. J Environ Radioact 142C:6267. doi:10.1016/j.jenvrad.2015.01.007

11. Beck HL, Bennett B (2002) Historical overview of atmospheric nuclear weapons
testing and estimates of fallout in the continental United States. Health Phys 82 (5):591608

247 12. Simon SL, Bouville A, Beck HL (2004) The geographic distribution of radionuclide
248 deposition across the continental US from atmospheric nuclear testing. J Environ
249 Radioact 74 (1-3):91-105. doi:10.1016/j.jenvrad.2004.01.023

13. Widner TE, Flack SM (2010) Characterization of the world's first nuclear explosion,
the Trinity test, as a source of public radiation exposure. Health Phys 98 (3):480-497.
doi:10.1097/HP.0b013e3181c18168

14. Lindahl P, Roos P, Eriksson M, Holm E (2004) Distribution of Np and Pu in Swedish
lichen samples (*Cladonia stellaris*) contaminated by atmospheric fallout. J Environ
Radioact 73:73-85

15. Zheng J, Tagami K, Watanabe Y, Uchida S, Aono T, Ishii N, Yoshida S, Kubota Y,
Fuma S, Ihara S (2012) Isotopic evidence of plutonium release into the environment from
the Fukushima DNPP accident. Sci Rep 2:304. doi:10.1038/srep00304

16. Schneider S, Walther C, Bister S, Schauer V, Christl M, Synal HA, Shozugawa K,
Steinhauser G (2013) Plutonium release from Fukushima Daiichi fosters the need for
more detailed investigations. Sci Rep 3:2988. doi:10.1038/srep02988

Purtyman WD, Peters RJ, Maes MN (1990) Plutonium deposition and distribution
from worldwide fallout in northern New Mexico and southern Colorado, Los Alamos
National Laboratory, LA-11794

18. Kelley JM, Bond LA, Beasley TM (1999) Global distribution of Pu isotopes and Np237. Sci Total Environ 237/238:483-500

- 19. Hicks HG, Barr DW (1984) Nevada test site fallout atom ratios: Pu-240/Pu-239 and
 Pu-241/Pu-239, Lewrence Livermore National Laboratory, UCRL-53499/1
- 269 20. Miller RL (1986) Under the cloud: the decades of nuclear testing. The Free Press: A
 270 division of Macmillan, Inc., USA
- 271 21. Hanson SK, Miller JL, Oldham WJ, Jr. unpublished work
- 272 22. Widner TE (2010) Final report of the Los Alamos historical document retrieval and
 273 assessment project, Centers for Disease Control and Prevention, LAHDRA
- 274 23. Thompson RC (1982) Neptunium The neglected actinide: a review of the
 biological and environmental literature. Radiation Research 90:1-32
- 24. Bunzl K, Kofuji H, Schimmack W, Tsumura A, Ueno K, Yamamoto M (1995)
 Residence times of global weapons testing fallout Np-237 in a grassland soil compared to
 Pu-239+240, Am-241, and Cs-137. Health Phys 68 (1):89-93
- 279 25. Hancock GJ, Tims SG, Fifield LK, Webster IT (2014) The release and persistence of 280 radioactive anthropogenic nuclides. *A Stratigraphic Basis for the Anthropocene*
- 281 Geological Society, London, Special Publications 395:265-281
- 282

283